

Dose dependent rhizospheric Ni toxicity evaluation: Membrane stability and antioxidant potential of *Vigna* species

Seema Mahmood¹, Shabnam Ishtiaq¹, Ghulam Yasin¹, and Ahsan Irshad^{2*}



ABSTRACT

Different concentrations of Ni are found in agricultural soil released from various environmental sources. The responses of plants grown in such soil-substratum vary accordingly depending upon the concentration of metal present and plant species. To assess the toxic effects of Ni and extent of plant defensive strategies, increasing Ni doses (50, 100, and 150 mg kg⁻¹) were used in sandy-loam soil-substratum at pH 7.9 to evaluate the performance of *Vigna* species (*V. cylindrica* [L.] Skeels, *V. mungo* [L.] Hepper, and *V. radiata* [L.] R. Wilczek). The experiment was conducted in a complete randomized design. Nickel stress was induced by adding various concentrations of Ni chloride in soil substratum. Malondialdehyde (MDA) and antioxidant levels were determined in roots and leaves. Escalating levels of Ni in soil resulted in an affirmative relationship between MDA with that of antioxidants. A dose-dependent increase in the activity of Superoxide Dismutases (SOD), Catalase (CAT), and peroxidases (POD) suggested the existence of a sequence response of these enzymes to scavenge oxidative stress in the roots. However, inadequate production of SOD and CAT appeared to be compensated by the enhanced activity of POD, which acted as potent quencher to reactive oxygen species (ROS) in leaves. At the most elevated Ni dose, SOD, CAT and POD activities were insufficient to counteract ROS generated that led to membrane damage manifested by elevated MDA levels. Nevertheless, SOD and CAT alleviated Ni toxicity in roots while SOD, CAT and POD acted in a concurrent manner to protect leaves from oxidative damage in *V. cylindrica*. The study clearly indicated a Ni dependent antioxidant enzymes defense system in *V. cylindrica*.

Key words: Antioxidant defense, catalasa, malondialdehyde (MDA), peroxidases, ROS.

¹Bahauddin Zakariya University, Institute of Pure and Applied Biology, Multan, Pakistan P.C. 60800.

²Institute of Crop Sciences, Chinese Academy of Agricultural Sciences, Beijing 100081, China.

*Corresponding author (ahsanaari@gmail.com).

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INTRODUCTION

Nickel (Ni) is a heavy metal naturally occurring in concentrations of 100 mg kg⁻¹ and 0.005 mg L⁻¹, in soil and water respectively (Chen et al., 2009; Bermudez et al., 2012). Soils derived from serpentine rocks are rich in Ni. Moreover, in the recent years, rapid industrialization and high anthropogenic pressures in the developing countries have caused 20-30 times increase of Ni concentration in the environment (Yusuf et al., 2012). The contamination of Ni in agricultural soils also results from industrial and municipal wastes, vehicular emissions, Ni fertilizers, electroplating, food industry and electric batteries (Tian et al., 2012; Sreekanth et al., 2013; Kaveriammal and Subramani, 2015). Consequently, the soil act as an ultimate sink of Ni then taken up by plants and enters the food webs (Ashok et al., 2010; Hussain et al., 2013; Mahar et al., 2015).

One of the important consequences of Ni induced phytotoxicity is the enhanced production of reactive oxygen species (ROS) which generates oxidative stress in plants (Kumar et al., 2012). The most damaging oxidative effect is lipid peroxidation of biological membranes, which results in the concomitant production of malondialdehyde (MDA). Thus, MDA levels have widely been utilized to estimate membrane stability potential in plants under metal stress (Dubey and Pandey, 2011).

The excessive ROS is scavenged by the production of both enzymatic and non enzymatic antioxidants to maintain normal metabolism and such strategies are universal features of plant defense to minimize oxidative damage (Ismail and Theodor, 2013). Oxidative damage could result when the balance between the detoxification of the ROS products and the antioxidative system is altered (Yusuf et al., 2011). The antioxidative activity may be of fundamental significance for the plants in their response against heavy metal stress including Ni (Fargasova, 2012). However, excess Ni has been found to reduce the activity of many cellular antioxidant enzymes and plant's capability to scavenge ROS, leading to ROS accumulation and finally oxidative stress in plants (Gill and Tuteja, 2010). Nevertheless, the plants have an innate ability to balance between generation and degradation of ROS (Gajewska et al., 2013). However, the activity of antioxidant enzymes may vary with the duration and type of stress treatment and within plant species (Gajewska and Sklodowska, 2007).

Superoxide dismutases (SODs) catalyze the dismutation of the superoxide radical (O₂⁻) into hydrogen peroxide (H₂O₂) and elemental oxygen (O₂). Among antioxidant enzymes, superoxide dismutase (SOD) is the major O₂⁻ scavenger and its enzymatic action results in H₂O₂ and O₂ formation. Catalase (CAT) and

several classes of peroxidases (POD) then scavenge the H_2O_2 produced. CAT dismutates H_2O_2 into H_2O and O_2 (Mishra et al., 2010) and finally POD decomposes H_2O_2 by oxidation of co-substrates such as phenolic compounds and/or antioxidants (Awasthi and Sinha, 2013). Thus, the activities of SOD, CAT, and POD have been well reported in plants species in response to heavy metal stress (Nadgorska-Socha et al., 2013; Ismail and Theodor, 2013).

Vigna species (*Vigna cylindrica* [L.] Skeels, *Vigna mungo* [L.] Hepper, and *Vigna radiata* [L.] R. Wilczek) are regarded as pulses/beans. They are important components of human diet all over the world because they contain high amount of easily digestible proteins. In addition, they are the best and least expensive sources of vegetable protein therefore serve as a substitute of meat protein for a significant proportion of population of the poor countries (Suneja et al., 2011; Nair et al., 2013). In Pakistan they are cultivated on 5% of the total cropped area (20 430 000 ha; PARC, 2012).

There exists little information on the formation of oxygen free radicals, membrane stability and antioxidative enzymes in response to Ni toxicity in *Vigna* species. Therefore, we investigated oxidative damage via lipid peroxidation and sequence of defensive responses represented by antioxidant enzymes (SOD, CAT, and POD) in tissues (roots and leaves) of *Vigna* species and their interspecific antioxidant potential in response to increasing doses of Ni for elaboration of mechanism for oxidative defense.

MATERIALS AND METHODS

Air-dried sandy loam soil substratum, pH 7.9, sieved through a 2 mm sieve was thoroughly mixed with nickel(II) chloride hexahydrate ($NiCl_2 \cdot 6H_2O$) (Merck, Darmstadt, Germany) to generate 50, 100 and 150 mg Ni kg^{-1} soil, while control plants were grown in soil without Ni. Since Ni salt was hexahydrated therefore actual concentration factor of Ni (4.05 g) was used to maintain Ni levels used in the study. The selected concentrations were used on the basis of Ni content reported in agriculture soils of Pakistan (sandy loam with pH 7.9) present in the vicinities of large cities and industries (Ishtiaq and Mahmood, 2011).

Sterilized seeds of *V. cylindrica* 'cp-386', *V. mungo* '6036-7' and *V. radiata* '97003' were obtained from Pulse Crop Division, Ayub Agriculture Research Institute, Faisalabad, Pakistan.

The experiment was arranged in a complete randomized manner. To simulate field conditions, plants were grown in a wire netting house under natural conditions (28 ± 5 °C, 12:12 h photoperiod, and 38% RH). Eight (pre-germinated) 6 d-old seedlings were transplanted into each of 36 earthen pots (height 45 cm and internal diameter 30 cm), which were filled with 8.0 kg soil substratum containing Ni. Seedlings were acclimatized for 1 wk then thinned out to four in each pot. Watering was carried out by gentle spraying using a spray gun to avoid leaching. Fifty days after germination, plants were harvested and leaves and roots were separated carefully for biochemical analyses.

Membrane stability was determined in terms of MDA concentrations in root and leaves by the method of Cakmak and Horst (1991) with minor modifications. Leaf and root samples (each 1.0 g) were homogenized in 3 cm^3 of 0.1% (w/v) trichloroacetic acid (TCA) solution. The homogenate was centrifuged at $20\,000 \times g$ for 15 min. Three cm^3 of 0.5% thiobarbituric acid (TBA) prepared in 20% TCA were added to 0.5 cm^3 supernatant. The mixture was heated at 95 °C in a shaking water bath for 50 min. The reaction was stopped by cooling the tubes in an ice water bath. Then the samples were centrifuged at $10\,000 \times g$ for 10 min, and the absorbance of the supernatant read at 532 and 600 nm. The MDA concentration was calculated as the difference in absorbance at 600 and 532 nm using extinction coefficient of $156\,mmol^{-1}\,cm^{-1}$.

For determination of antioxidants enzymes, fresh leaves and roots (0.5 g) were ground using a pestle and mortar in 5 cm^3 of cooled 50 mM phosphate buffer (pH 7.8). After filtration through cheesecloth, the homogenate was centrifuged at $15\,000 \times g$ for 20 min at 4 °C and the supernatant was used in further assay for superoxide dismutase (SOD) activity (Giannopolitis and Ries, 1977), peroxidase and catalase activity (Chance and Maehly, 1955) by recording absorbance at 560, 240, and 470 nm respectively.

SOD activity inhibits the photochemical reduction of nitro blue tetrazolium (NBT) at 560 nm. The monitoring of this inhibition is used to assay SOD activity. The reaction mixture was prepared by taking 0.050 cm^3 enzyme extract and adding 1 cm^3 NBT (50 μM), 0.5 cm^3 methionine (13 mM), 1 cm^3 riboflavin (1.3 μM), 0.95 cm^3 (50 mM) phosphate buffer and 0.5 cm^3 EDTA (75 mM). This reaction was started by keeping reaction solution under 30 W fluorescent lamp illuminations and turning the fluorescent lamp on. The reaction stopped when the lamp turned off 5 min later. The NBT photo reduction produced blue formazan, which was used to measure the increase in absorbance at 560 nm. The same reaction mixtures without enzyme extract in dark were used as blank. The SOD activity was determined and expressed as SOD I $\mu min^{-1}\,mg^{-1}$ protein (Giannopolitis and Ries, 1977) (1 unit is the amount of SOD which inhibits the rate of increase in absorbance due to NBT-diformazan formation by 50%).

Catalase (CAT) activity assayed by the decomposition of H_2O_2 and change in absorbance due to H_2O_2 was observed every 30 s for 5 min at 240 nm using a UV-visible spectrophotometer (Hitachi 2000, Japan). Reaction mixture for CAT contained 0.9 cm^3 H_2O_2 (5.9 mM) and 2 cm^3 phosphate buffer (50 mM). Reaction was started by adding 0.1 cm^3 enzyme extract to the reaction mixture. The CAT activity was expressed as $\mu mol\,H_2O_2\,min^{-1}\,mg^{-1}$ protein (Chance and Maehly, 1955).

Peroxidase (POD) activity assayed by guaiacol oxidation and defined as 0.01 absorbance change $min^{-1}\,mg^{-1}$ protein. The reaction mixture was prepared by adding 0.4 cm^3 guaiacol (20 mM), 0.5 cm^3 H_2O_2 (40 mM) and 2 cm^3 phosphate (50 mM) in 0.1 cm^3 enzyme extract. The change in absorbance at 470 nm of the reaction mixture was observed every 20 s

up to 5 min. The POD activity expressed as $\text{mmol min}^{-1} \text{mg}^{-1}$ protein (Chance and Maehly, 1955).

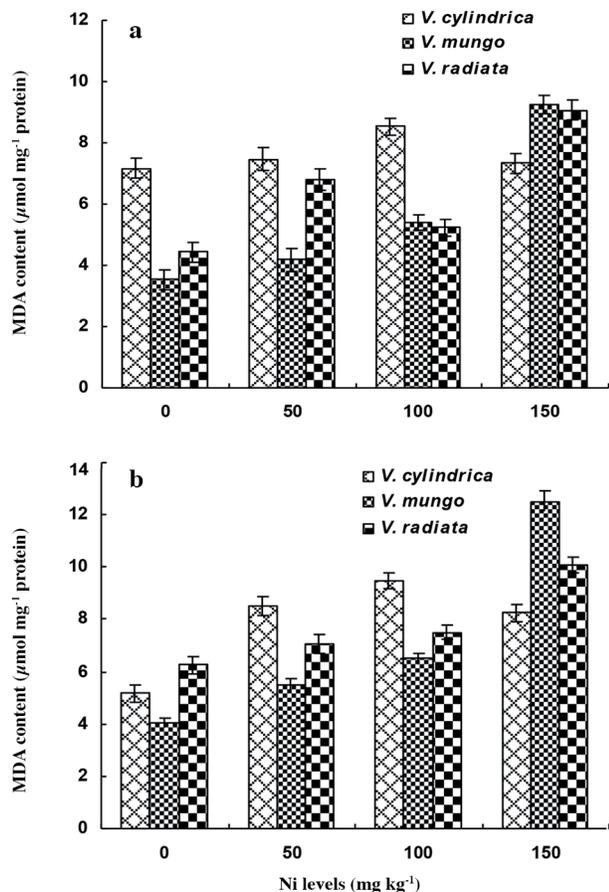
Statistical analysis

The statistical analysis was performed on data replicated thrice. Each parameter was subjected to a three-way ANOVA using COSTAT (1998-2004 Cohort Software) in order to determine significant differences among Ni levels, species and their tissues for the biochemical analyses carried out. Significant Differences (LSD) between means for each main factor were calculated by employing a multiple range test following Snedecor and Cochran (1989).

RESULTS

In this study, the formation of MDA was considered as a measure of lipid peroxidation in leaves (Figure 1a) and roots (Figure 1b) of three *Vigna* species. It is evident from data that elevation of MDA in both tissues (leaves and roots) of all the species occurred in response to Ni toxicity in a concentration dependent manner. The MDA contents were greater in roots of the species being the highest ($12.5 \text{ mmol}^{-1} \text{cm}^{-1}$) for *V. mungo* at $150 \text{ mg Ni kg}^{-1}$ as compared with leaves ($9.25 \text{ mmol}^{-1} \text{cm}^{-1}$) followed by *V. radiata*. The extent of

Figure 1. Malondialdehyde (MDA) content in leaves (a) and roots (b) of three *Vigna* species in response to varying Ni levels. Results are the mean of three replicates \pm SE.



elevation of MDA in *V. mungo* was 210% and 162% in roots and leaves, respectively at $150 \text{ mg Ni kg}^{-1}$. A steady increase in leaf MDA content was also observed in *V. cylindrica* at 50 and 100 mg kg^{-1} and showed 4% and 19% rise, respectively. The most significant increase (83%) in MDA was noticed in roots of *V. cylindrica* at $100 \text{ mg Ni kg}^{-1}$ when compared to its respective control. The species had significantly ($P < 0.001$) variable responses to Ni application and likewise the MDA content in the tissues also varied markedly ($P < 0.001$) as shown in Table 1.

There has been observed a progressive increase in SOD activity in leaves (Figure 2a) and roots (Figure 2b) of the species after exposure to Ni levels. The change in SOD activity in roots was more pronounced in *V. cylindrica* ($174.67 \mu \text{ mg}^{-1} \text{ protein}$) followed by *V. radiata* ($150.33 \mu \text{ mg}^{-1} \text{ protein}$) after exposure to higher Ni concentrations. SOD production in leaves was slightly lower at 50 mg kg^{-1} for all *Vigna* species. A consistently lower SOD activity in both tissues of *V. mungo* was observed at the most elevated Ni level. The results (Table 1) also depicted differential production of SOD in the tissues ($P < 0.001$) marked contrast between species ($P < 0.001$).

A consistent increase in CAT was observed in leaves of Ni-treated plants of *Vigna* species but stronger CAT activity was observed in *V. cylindrica* and *V. radiata* at $150 \text{ mg Ni kg}^{-1}$ (Figure 2c). However, at 150 mg kg^{-1} the degree of increase in *V. cylindrica* was up to 162% when compared with control. The most elevated Ni level induced 78% rise in leaf CAT in *V. radiata* followed by 44% in *V. mungo*. In roots, more profound change (139% rise) in the CAT activity was observed in *V. radiata* in response to the most elevated Ni level (Figure 2d).

Table 1. ANOVA of the activity of superoxide dismutase (SOD), catalase (CAT) and peroxidases (POD) in leaves and roots of three *Vigna* species in response to varying Ni levels.

	MDA	SOD	CAT	POD
Plant tissues				
Significant range	***	***	***	***
LSD	0.135	2.149	0.675	0.01
Leaves	7.53a	138.47a	12.33a	0.99a
Roots	6.45b	64.25b	8.24b	0.17b
Species				
Significant range	***	***	***	***
LSD	0.110	2.631	0.826	0.01
<i>V. cylindrica</i>	8.50a	114.29a	9.53a	0.71a
<i>V. mungo</i>	7.09b	102.42b	12.75b	0.56b
<i>V. radiata</i>	5.37c	87.37c	8.58c	0.46c
Ni levels, mg kg^{-1}				
Significant range	***	***	***	***
LSD	0.156	3.039	0.95	0.014
Control	7.42a	114.44a	12.17a	0.69a
50	6.93b	113.5a	12.02a	0.61b
100	6.84b	89b	10.05b	0.56c
150	6.76b	88.5b	6.91c	0.45d

***Significant at the 0.001 probability level.

Each value represents mean of three replicates. Means were compared using ANOVA. Data followed by different letters (a-d) in respective columns are significantly different at $P < 0.05$.

LSD: Least significant difference.

Figure 2. The activities of superoxide dismutase (SOD, a) catalase (CAT, c), and peroxidases (POD, e), in leaves and SOD (b), CAT (d), and POD (f) in roots of three *Vigna* species in response to varying Ni levels. Results are the mean of three replicates \pm SE.

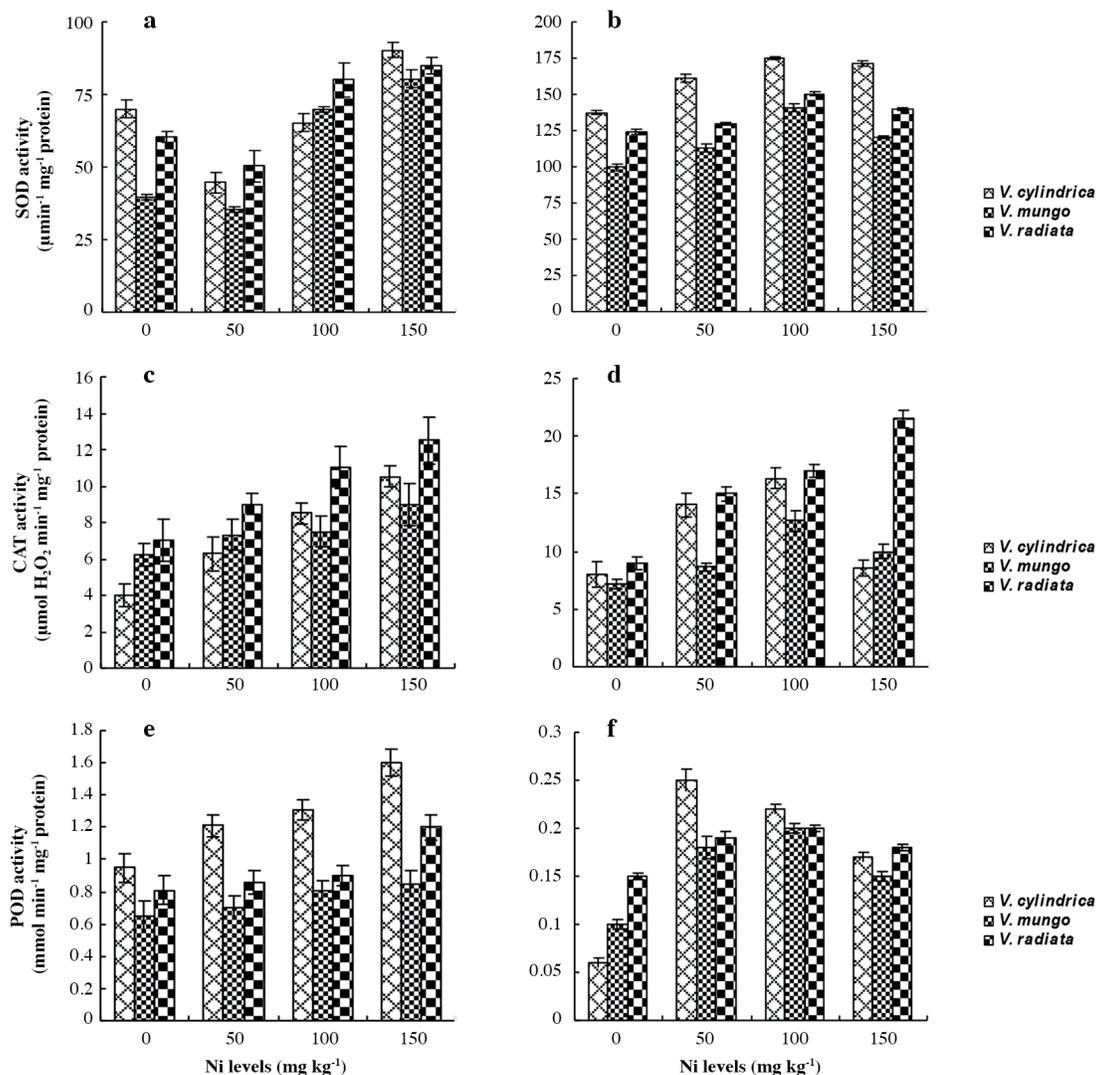


Figure 2e depicted that the *Vigna* species had a supreme approach of POD activity in leaves in response to Ni applications. At 150 mg Ni kg⁻¹, POD activity peaked (68% rise) in leaves of *V. cylindrica* followed by *V. radiata* and *V. mungo*, which showed 48% and 31% elevation, respectively than their control plants. A steady increase in POD activity was also noticed in roots of *Vigna* species with increasing Ni levels (Figure 2f). POD activity changed drastically in the roots of *V. cylindrica* by 183% at 150 mg kg⁻¹ Ni. However, no marked increase in POD activity became evident in roots of *V. radiata*. In general, POD activity enhanced greatly ($P < 0.001$) leaves than roots (Table 1).

DISCUSSION

Oxidative stress in higher plants induced by heavy metals leads to the generation of superoxide radical (O_2^-), hydrogen peroxide (H_2O_2), hydroxyl radical (OH^\cdot), and singlet oxygen ($1O_2$), collectively termed ROS (Kazemi et al., 2010). ROS

can damage all classes of biological molecules such as nucleic acids, proteins, lipids, and amino acids (Hussain et al., 2013) causing irreparable tissue damage and ultimately cell death (Awasthi and Sinha, 2013).

Toxic effects of Ni are probably exerted through free radical generation, which results in the enhanced production of MDA (a lipid peroxidation product) in *Vigna* species. The results of this study indicated rapid and continuous lipid peroxidation stimulated by Ni and assessed as MDA concentration (Table 1). The fast rise of membrane lipid peroxidation could be due to direct contact of roots with the metal (Dubey and Pandey, 2011; Singh et al., 2013). Several other studies demonstrated that Ni toxicity contributed enhanced generation of ROS, thus peroxidative damage to membrane lipids is highly likely (Das et al., 2008; Gill and Tuteja, 2010). Moreover, Ni-induced depletion of some low molecular weight proteins (GSH), may lead to damage of membrane, thus, may contribute to the induction of oxidative stress in plants (Sreekanth et al., 2013).

Heavy metals enters many types of cells and under physiological conditions can be reduced by hydrogen peroxide (HO), glutathione (GSH) reductase, ascorbic acid, and GSH to produce reactive intermediates, including thyl radicals, hydroxyl radicals, etc. Any of these species could attack DNA, proteins, and membrane lipids, thereby disrupting cellular integrity and functions.

The result indicated that more SOD and CAT activities was noticed in roots of *V. cylindrica* followed by *V. radiata* under high Ni doses, while insufficient responses of SOD and CAT to scavenge oxidative stress in leaves of all the *Vigna* species were balanced by the enhanced activity of POD to protect leaves from oxidative damage (Figure 2). A protective mechanism to minimize oxidative damage generated by ROS initiates which comprises the production of important antioxidant enzymes including SOD, CAT, and POD (Lu et al., 2010; Idrees et al., 2013). Exposure of plants to Ni at low concentrations (0.05 mM) for short times has been shown to increase the activity of the antioxidant enzymes of the ascorbate-glutathione cycle (Kachout et al., 2009), which protect plant cells against free radicals by removing (or scavenging) of ROS (Ismail and Theodor, 2013).

The minimum activity of SOD in *V. mungo* at 150 mg kg⁻¹ (Table 1) indicated that the least adaptive nature of *V. mungo*. While, SOD activity in roots of *V. cylindrica* and *V. radiata* peaked at higher Ni concentrations suggesting that this increase in SOD has better protection against oxidant damage. This increase in the concentration of SOD in *V. cylindrica* and *V. radiata* under high Ni doses might be due to the induction of genes of SOD by superoxide mediated signal transduction, which causes de novo synthesis of enzyme proteins (Tewari et al., 2008) and thus causing more superoxide generation. The enzyme concentrations were more increased under high Ni supplements and thus the higher concentration of antioxidant enzymes were involved in the protective mechanisms adapted by plants. This can be correlated to the increased stress tolerance (Awasthi and Sinha, 2013). A rise in SOD concentration in the observe study was found to be more pronounced in roots as compared to leaves (Table 1). Such increase in the activity of these enzymes has also been reported with a direct contact of root with the metal applied for treatments and the extent of increase varied with the metal concentration and the plant species (Duman and Ozturk, 2010; Hussain et al., 2013). Decrease in SOD activity might be due to the competitive uptake of Ni against the other metals ions being the co- factors used by respective forms of this enzyme (Pandey and Sharma, 2002) as well as to its inactivation by H₂O₂ or other ROS (Lu et al., 2010).

Some forms of antioxidant could be rendered inadequate in the species because of the higher Ni doses induced alterations in the activities of alternate enzymes of antioxidant defense (SOD, CAT). The oxidative damage to membrane in Ni stress *V. mungo* could be due to accumulation of H₂O₂ and a consequent of enhanced activity of SOD and inhibition of CAT. The increase in SOD and decrease in CAT activity in response to excess Ni supply has been widely reported by other workers also (Pandey et al., 2009).

POD is a stress marker enzyme located in the cell wall, vacuole and extracellular spaces utilize H₂O₂ to generate phenoxy compounds that polymerize to produce lignin (Gajewska and Sklodowska, 2010) and is involved in strengthening of the cell wall by promoting lignin biosynthesis (Chaoui and El-Ferjani, 2005). Thus, the stronger cell wall itself acts as a physical barrier against toxic metal ions (Fargasova, 2012). The substrate affinity of POD is higher than that of CAT (Mhamdi et al., 2010). Therefore, the results indicated that *V. cylindrica* might efficiently avoid heavy metal by showing its capacity to produce more POD. The POD activity of both the roots and the leaves of *V. cylindrica*, at increasing doses of Ni, suggested that POD can affect H₂O₂ detoxification. The reduction in CAT at high Ni concentration might be due to the condition that during stress condition the tetrameric CAT molecules may degrade into monomeric CAT subunits, which act like POD (Panda and Khan, 2003). Moreover, Gajewska and Sklodowska (2010) described that peroxidase-catalyzed reactions, including lignification and cross-links formation, result in decreased cell wall plasticity and consequently in its restricted elongation. POD itself might be utilized in consuming H₂O₂ to generate phenoxy compounds that are polymerized to produce cell wall components such as lignans (Kovacik et al., 2009). Increased POD activity shown in our results is correlated with Ni stress suggesting it to be an intrinsic defense tool and can be related with the release of peroxidase localized in the cell walls (Chen et al., 2000). These results are in consonance with earlier reports which also indicated that lower response of SOD activity compensated by the increased activities of CAT and POD (Maheshwari and Dubey, 2009; Duman and Ozturk, 2010; Ismail and Theodor, 2013).

CONCLUSIONS

It was concluded from this study that applied Ni implication (100-150 mg kg⁻¹) induced lipid peroxidation leading to membrane damage as evident from enhanced malondialdehyde concentration. This oxidative damage was more profound in roots than leaves due to direct contact of membrane root with the metal. Under high doses of Ni, superoxide dismutase and catalase appeared to be compensated by the enhanced activity of peroxidases as potent quencher to reactive oxygen species in leaves. Among the species, *Vigna mungo* was most sensitive and *V. cylindrica* was the most tolerant under high Ni doses.

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REFERENCES

- Ashok, K., B.S. Bisht, K. Manish, and K. Lalit. 2010. Effects of Ni and Zn on growth of *Vigna mungo*, *Vigna radiata* and *Glycine max*. International Journal of Pharmacy and Biological Sciences 2:1-7.
- Awasthi, K., and P. Sinha. 2013. Nickel stress induced antioxidant defence system in sponge gourd (*Luffa cylindrica*). Journal of Plant Physiology & Pathology 1:1.
- Bermudez, G.M.A., R. Jasan, R. Pla, and M.L. Pignata. 2012. Heavy metals and trace elements in atmospheric fall-out: Their relationship with topsoil and wheat element composition. Journal of Hazardous Materials 213-214:447-456.
- Cakmak, I., and W.J. Horst. 1991. Effect of aluminum on lipid peroxidation, superoxide dismutase, catalase and peroxidase activities in root tips of soybean (*Glycine max*). Physiologia Plantarum 83:463-468.
- Chance, B., and A.C. Maehly. 1955. Assay of catalases and peroxidases. Methods in Enzymology 2:764-775.
- Chaoui, A., and E. El-Ferjani. 2005. Effect of cadmium and copper on antioxidant capacities, lignification and auxin degradation in leaves of pea (*Pisum sativum* L.) seedlings. Comptes Rendus Biologies 328:23-31.
- Chen, C., D. Huang, and J. Liu. 2009. Functions and toxicity of nickel in plants: Recent advances and future prospects. Clean Soil Air Water 37:304-313.
- Chen, L.M., C.C. Lin, and C.H. Kao. 2000. Copper toxicity in rice seedlings: changes in antioxidative enzyme activities, H₂O₂ level, and cell wall peroxidase activity in roots. Botanical Bulletin of Academia Sinica 41:99-103.
- Das, K.K., S.N. Das, and S.A. Dhundasi. 2008. Nickel, its adverse health effects and oxidative stress. Indian Journal of Medical Research 128:412-425.
- Dubey, D., and A. Pandey. 2011. Effect of nickel (Ni) on chlorophyll, lipid peroxidation and antioxidant enzymes activities in black gram (*Vigna mungo*) leaves. International Journal of Science and Nature 2:395-401.
- Duman, F., and F. Ozturk. 2010. Nickel accumulation and its effect on biomass, protein content and antioxidative enzymes in roots and leaves of watercress (*Nasturtium officinale* R. Br.) Journal of Environmental Sciences 22:526-532.
- Fargasova, A. 2012. Plants as models for chromium and nickel risk assessment. Ecotoxicology 21:1476-1483.
- Gajewska, E., and M. Sklodowska. 2007. Effect of nickel on ROS content and antioxidative enzyme activities in wheat leaves. BioMetals 20:27-36.
- Gajewska, E., and M. Sklodowska. 2010. Differential effect of equal copper, cadmium and nickel concentration on biochemical reactions in wheat seedlings. Ecotoxicology and Environmental Safety 73:996-1003.
- Gajewska, E., E. Niewiadomsk, K. Tokarz, M. Slab, and M. Sklodowsk. 2013. Nickel-induced changes in carbon metabolism in wheat shoots. Journal of Plant Physiology 170:369-377.
- Giannopolitis, C.N., and S.K. Ries. 1977. Superoxide dismutase I. Occurrence in higher plants. Journal of Plant Physiology 59:309-314.
- Gill, S.S., and N. Tuteja. 2010. Reactive oxygen species and antioxidant machinery in abiotic stress tolerance in crop plants. Plant Physiology and Biochemistry 48:909-930.
- Hussain, M.B., S. Ali, A.H.S. Azam, M.A. Farooq, B. Ali, S.A. Bharwana, et al. 2013. Morphological, physiological and biochemical responses of plants to nickel stress: A review. African Journal of Agricultural Research 8:1596-1602.
- Idrees, M., M. Naeem, T. Aftab, M. Masroor, A. Khan, and Moinuddin. 2013. Salicylic acid restrains nickel toxicity, improves antioxidant defence system and enhances the production of anticancer alkaloids in *Catharanthus roseus* (L.) Journal of Hazardous Materials 252-253:367-374.
- Ishtiaq, S., and S. Mahmood. 2011. Phytotoxicity of nickel and its accumulation in tissues of three *Vigna* species at their early growth stages. Journal of Applied Botany and Food Quality 84:223-228.
- Ismail, M.A., and P.A. Theodor. 2013. Effects of zinc and nickel on antioxidative enzyme activities of hairy roots of *Brassica juncea* L. Czern (Indian mustard). International Journal of Advanced Biotechnology and Research 3:53-60.
- Kachout, S.S., A.B. Mansoura, J.C. Leclerc, R. Mechergui, M.N. Rejeb, and Z. Oueghi. 2009. Effects of heavy metals on antioxidant activities of *Atriplex hortensis* and *A. rosea*. Journal of Applied Botany and Food Quality 83:37-43.
- Kaveriammal, S., and A. Subramani. 2015. Variation in seed germination and early growth of groundnut (*Arachis hypogaea* L.) under nickel treatments. International Journal of Environment and Bioenergy 10:47-53.
- Kazemi, N., R.A. Khavari-Nejad, H. Fahimi, S. Saadatmand, and T. Nejad-Sattari. 2010. Effect of exogenous salicylic acid and nitric oxide on lipid peroxidation and antioxidant enzyme activities in leaves of *Brassica napus* L. under nickel stress. Scientia Horticulturae 126:402-407.
- Kovacik, J., B. Klejdus, J. Kadukova, and M. Backor. 2009. Physiology of *Matricaria chamomilla* exposed to nickel excess. Ecotoxicology and Environmental Safety 72:603-609.
- Kumar, H., D. Sharma, and V. Kumar. 2012. Nickel-induced oxidative stress and role of antioxidant defense in Barley roots and leaves. International Journal of Environmental Biology 2:121-128.
- Lu, M.Y., X.R. Li, M.Z. He, Z.N. Wang, and H.J. Tan. 2010. Nickel effects on growth and antioxidative enzymes activities in desert plant *Zygophyllum xanthoxylon* (Bunge). Sciences in Cold and Arid Regions 2:0436-0444.
- Mahar, A., P. Wang, R. Li, and Z. Zhang. 2015. Immobilization of lead and cadmium in contaminated soil using amendments: A review. Pedosphere 25:555-568.
- Maheshwari, R., and R.S. Dubey. 2009. Nickel-induced oxidative stress and the role of antioxidant defence in rice seedlings. Plant Growth Regulation 59:37-49.
- Mhamdi, A., G. Queval, S. Chaouch, S. Vanderauwera, F. Van Breusegem, and G. Noctor. 2010. Catalase function in plants: a focus on Arabidopsis mutants as stress-mimic models. Journal of Experimental Botany 15:4197-4220.
- Mishra, S., D. Panjwani, B. Mishra, and P.N. Sharma. 2010. Effect of excess nickel on induction of oxidative stress in *Zea mays* L. plants grown in solution culture. International Journal of Toxicological and Pharmacological Research 2:10-15.
- Nadgorska-Socha, A., A. Kafel, M. Kandziora-Ciupa, J. Gospodarek, and A. Zawisza-Raszka. 2013. Accumulation of heavy metals and antioxidant responses in *Vicia faba* plants grown on monometallic contaminated soil. Environmental Science and Pollution Research International 20:1124-1134.
- Nair, R.M., R.Y. Yang, W.J. Easdown, D. Thavarajah, P. Thavarajah, J.A. Hughes, and J.D.H. Keatinge. 2013. Biofortification of mungbean (*Vigna radiata*) as a whole food to enhance human health. Journal of the Science of Food and Agriculture 93:1805-1813.
- Panda, S.K., and M.H. Khan. 2003. Antioxidant efficiency in *Oryza sativa* leaves under heavy metal toxicity. Journal of Plant Biology 30:23-29.

- Pandey, N., and C.P. Sharma. 2002. Effect of heavy metals Co^{2+} , Ni^{2+} and Cd^{2+} on growth and metabolism of cabbage. *Plant Science* 163:753-758.
- Pandey, N., G.C. Pathak, D.K. Pandey, and R.L. Pandey. 2009. Heavy metals, Co, Ni, Cu, Zn and Cd, produce oxidative damage and evoke differential antioxidant responses in spinach. *Brazilian Journal of Plant Physiology* 21:103-111.
- PARC. 2012. National Coordinated Pulses Programme. Pakistan Agriculture Research Council (PARC), Islamabad, Pakistan. Available at <http://www.parc.gov.pk/index.php/en/csi/137-narc/crop-sciences-institute/712-national-coordinated-pulses-programme> (accessed January 2016).
- Singh, G., R.K. Agnihotri, D.K. Singh, and R. Sharma. 2013. Effect of Pb and Ni on root development and biomass production of black gram (*Vigna mungo* L.): overcoming through exogenous nitrogen application. *International Journal of Agriculture and Crop Sciences* 5:1410-1417.
- Snedecor, G.W., and W.G. Cochran. 1989. *Statistical methods*. 8th ed. 503 p. Iowa State University Press, Ames, Iowa, USA.
- Sreekanth, T.V.M., P.C. Nagajyothi, K.D. Lee, and T.N.V.K.V. Prasad. 2013. Occurrence, physiological responses and toxicity of nickel in plants. *International Journal of Environmental Science and Technology* 10:1129-1140.
- Suneja, Y., S. Kaur, A.K. Gupta, and N. Kau. 2011. Levels of nutritional constituents and antinutritional factors in black gram (*Vigna mungo* L. Hepper). *Food Research International* 44:621-628.
- Tewari, A., R. Singh, N.K. Singh, and U.N. Rai. 2008. Amelioration of municipal sludge by *Pistia stratiotes* L.: Role of antioxidant enzymes in detoxification of metals. *Bioresource Technology* 99:8715-8721.
- Tian, H.Z., L. Lu, K. Cheng, J.M. Hao, D. Zhao, Y. Wang, et al. 2012. Anthropogenic atmospheric nickel emissions and its distribution characteristics in China. *Science of the Total Environment* 1417-418:148-157.
- Yusuf, M., Q. Fariduddin, S. Hayat, and A. Ahmad. 2011. Nickel: An overview of uptake, essentiality and toxicity in plants. *Bulletin of Environmental Contamination and Toxicology* 86:1-17.
- Yusuf, M., Q. Fariduddin, and A. Ahmad. 2012. 24-Epibrassinolide modulates growth, nodulation, antioxidant system, and osmolyte in tolerant and sensitive varieties of *Vigna radiata* under different levels of nickel: A shotgun approach. *Plant Physiology* 57:143-153.