# Tropical Journal of Pharmaceutical Research February 2016; 15 (2): 371-375

**ISSN:** 1596-5996 (print); 1596-9827 (electronic)

© Pharmacotherapy Group, Faculty of Pharmacy, University of Benin, Benin City, 300001 Nigeria.

All rights reserved.

Available online at http://www.tjpr.org http://dx.doi.org/10.4314/tjpr.v15i2.21

# **Original Research Article**

# Mycosynthesis of Silver Nanoparticles from *Candida albicans* and its Antibacterial Activity against *Escherichia coli* and *Staphylococcus aureus*

# Ghasem Rahimi, Fahimeh Alizadeh\* and Alireza Khodavandi

Department of Microbiology, Yasooj Branch, Islamic Azad University, Yasooj, Iran

\*For correspondence: Email: falizadeh@iauyasooj.ac.ir; Tel: +987433313930 Fax: +987433312200

Received: 20 July 2015 Revised accepted: 30 January 2016

#### Abstract

**Purpose:** To produce and characterize silver nanoparticles using Candida albicans and evaluate its antibacterial properties.

**Methods:** Extracellular silver nanoparticles were biosynthesized using C. albicans. The biomass obtained from cultures of C. albicans was used to synthesize silver nanoparticles in 1.5 mM silver nitrate solution. Characterization of the biosynthesized nanoparticles was carried out using ultraviolet (UV)-visible spectrometry and scanning electron microscopy (SEM). The antibacterial properties of the nanoparticles were determined by agar disc diffusion method against Escherichia coli and Staphylococcus aureus.

**Results:** Incubation of C. albicans with silver nitrate solution produced silver nanoparticles after 2-4 h as evidenced by change in the color of the reaction mixture. The formation of silver nanoparticles was confirmed by UV-visible spectroscopy which showed absorption peak between 300-800 nm, being the characteristic wavelength range of silver nanoparticles. SEM revealed the varying morphology of the nanoparticles which had a size range of 20-80 nm. Furthermore, the nanoparticles showed significant antimicrobial activity (p < 0.05).

**Conclusion:** The biosynthesized silver nanoparticles hold some promise for various industrial applications, including drug delivery.

**Keywords:** Antibacterial activity, Candida albicans, Escherichia coli, Mycosynthesis, Silver nanoparticles, Staphylococcus aureus

Tropical Journal of Pharmaceutical Research is indexed by Science Citation Index (SciSearch), Scopus, International Pharmaceutical Abstract, Chemical Abstracts, Embase, Index Copernicus, EBSCO, African Index Medicus, JournalSeek, Journal Citation Reports/Science Edition, Directory of Open Access Journals (DOAJ), African Journal Online, Bioline International, Open-J-Gate and Pharmacy Abstracts

#### INTRODUCTION

The use of nanotechnology in medicine is set to spread rapidly. In the last decades, the treatment of bacterial infections has become increasingly complicated due to their resistance mechanisms. This has resulted in severe debilitating and even life-threatening infections; thus this situation is driving the search for novel antibacterial agents [1].

Metal nanoparticles have proved to be effective against microorganisms. The broad-spectrum antimicrobial properties of silver encourage its widespread use [1,2]. Silver nanoparticles are of particular interest due to their distinctive physical, chemical and biological properties [3,4]. Silver nanoparticles have proved to be most effective metallic nanoparticles as they have potent antimicrobial efficacy against various pathogenic microorganisms including bacteria such as Escherichia coli, Staphylococcus aureus,

Staphylococcus epidermis, Leuconostoc mesenteroides. Bacillus subtilis. Klebsiella and Klebsiella Silver mobilis, pneumonia. nanoparticles are more effective against microorganisms due to high surface area to volume ratio so that a large proportion of silver atoms are in direct contact with their environment. The antibacterial activity of silver nanoparticles against Gram negative and Gram positive bacteria have been established, but the mechanism of action of silver nanoparticles has not been completely understood [2,3,5-8].

Recently, researchers found that microorganisms have been explored as potential ecofriendly nanofactories for the synthesis of silver nanoparticles. Biosynthesis of nanoparticles has received considerable attention due to better control over size distribution of nanoparticles and the environmental toxicity [5,9,10]. The potential application of microorganisms in nanoparticles synthesis has been explored that can be applied to many fields and more specifically to biomedicine. Thus, significant efforts have been directed towards finding an inexpensive and environment-friendly method for nanoparticle synthesis. Several lines of evidence demonstrated the biosynthesis of extracellular silver nanoparticles utilizing many ubiquitous fungal species. Fungi especially yeast present a suitable option for large-scale nanoproduction. The mechanism of nanoparticle production using fungi is different; fungi secrete large amounts of enzymes which are used to reduce silver ions to metal nanoparticles mycosynthesis of [3,11,12]. The silver nanoparticles was carried out from Candida albicans and their antibacterial activity against E. coli and S. aureus was evaluated.

## **EXPERIMENTAL**

## Microorganisms

In this study 15 clinical isolates of *Candida albicans* were isolated and identified from immunocompromised patients. Standard strain of *C. albicans* ATCC 14053, *Escherichia coli* ATCC 39403 and *Staphylococcus aureus* ATCC 25923 were employed.

#### Biosynthesis of silver nanoparticles

The characterized isolates of *C. albicans* and *C. albicans* ATCC 14053 were plated out on Sabouraud Dextrose Agar (SDA, Merck, Germany) containing 3.5 mM silver nitrate and incubated at 37 °C for 24 h. The colonies

obtained were inoculated in 250 ml Sabouraud Dextrose Broth (SDB, Merck, Germany) and incubated at 37 °C on shaking incubator at 200 rpm for 24 h. After incubation the cells were separated by centrifugation at 6,000 rpm for 15 min and washed 3 times with 50 mM phosphate buffer (pH 7.0). Two g of wet biomass was collected and then added in 1.5 mM 50 ml silver nitrate solution and incubated at 37 °C on shaking incubator at 250 rpm in dark for 48 h. The silver nitrate reduced to silver nanoparticles shows change in color of silver nitrate from colorless to brown was harvested by centrifugation at 6,000 rpm for 6 min. The production of silver nanoparticles were analyzed by UV-visible spectrophotometric analysis and subjected to scanning electron microscopy (SEM) [3].

## Characterization of silver nanoparticles

The synthesis of silver nanoparticles was monitored by UV-visible spectroscopy (Shimadzu, UV Pharma spec 1700 with a resolution of 0.72 nm) by recording the spectra between 300 - 800 nm. The cell biomass of silver nitrate at time zero served as control. The produced silver nanoparticles were subjected to SEM (JEOL JSM 6360, Japan) and the images of nanoparticles were taken. The exact size and particle size distributions of silver nanoparticles were also determined by particle size analysis.

#### Antibacterial activity of silver nanoparticles

Silver nanoparticles synthesized from *C. albicans* were tested for antibacterial activity against pathogenic microorganisms *E. coli* ATCC 39403 and *S. aureus* ATCC 25923. The antibacterial activity of silver nanoparticles was tested by the agar disc diffusion method according to the Clinical and Laboratory Standards Institute protocol (M02 – A11) for bacteria with some slight modification. Both bacteria were cultured on Nutrient Agar (NA, Merck, Germany) and then passaged twice to ensure viability and purity.

Stock inoculum suspensions of the *E. coli* and *S. aureus* were prepared by picking colonies from 24 h cultures grown on NA at 37 °C and suspending in 5 ml of sterile saline (0.85 %) v/v NaCl. The suspension was adjusted with the aid of spectrophotometry at a wavelength of 625 nm to achieve turbidity equivalent to 0.5 McFarland standard, i.e., approximately  $(1-2) \times 10^8$  colony-forming units /ml). The inocula were spread on Mueller Hinton Agar using a sterile cotton swab. The culture plates were kept to dry at room temperature for 15 min. For the next step 0, 10, 20 and 50  $\mu$ l of the sterile silver nanoparticles

were impregnated on separate paper disks and put on the surface of cultured media. Tetracycline was used as positive control. Eventually the cultured plates were incubated at 37 °C for 24 h and then the diameter of zone of inhibition was measured. Three technical replicates were performed in each test and experiment was repeated at least two times [13].

#### Statistical analysis

Statistical analysis was done using SPSS software (version 21, SPSS Inc, Chicago, IL, USA). Normality and ANOVA were used to compare results. P < 0.05 was considered significant.

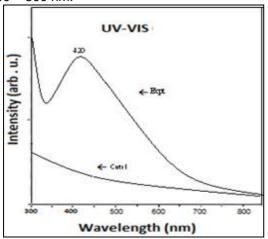
## **RESULTS**

## **Biosynthesized silver nanoparticles**

Silver nanoparticles were synthesized from silver nitrate by C. albicans. The reaction started within 2-4 h and the color of the solution turned to yellowish brown, indicating the synthesis of silver nanoparticles by C. albicans.

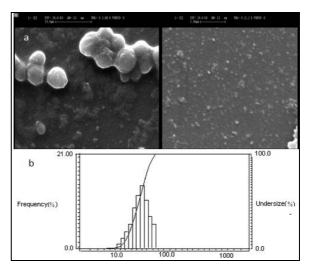
#### Characteristics of silver nanoparticles

The formation of silver nanoparticles by *C. albicans* was further determined using the UV–visible spectroscopy, which was shown on the surface plasmon resonance bands. Figure 1 demonstrated that the silver nanoparticles peak was observed at 300 – 800 nm and the silver nanoparticles contribute to the absorption bands at 420 nm in the UV–visible spectra. UV–Vis spectra of aqueous silver nitrate solution (control) when monitored separately did not show any at 300 – 800 nm.



**Figure 1:** The UV-visible spectra curve of silver nanoparticles synthesized by *Candida albicans* 

The SEM images and their corresponding particle size distributions of silver nanoparticles are shown in Figure 2. The SEM images and their size distributions revealed that, the mean diameters and standard deviation of silver nanoparticles were about 20 – 80 nm with the different morphologies. Electron microscopy data indicate that the extracellular particles produced.



**Figure 2.** a: Scanning electron microscopy image and b: the particle size distribution for silver nanoparticles synthesized by *Candida albicans* 

## Antibacterial activity

The antibacterial activity of synthesized silver nanoparticles by cell biomass of C. albicans was tested against pathogenic microorganisms E. coli and S. aureus. The synthesized silver nanoparticles efficiently inhibited the growth of Gram negative and Gram positive bacteria. Our results demonstrated that the biosynthesized silver nanoparticles had more antimicrobial activity against S. aureus compared to E. coli. The zones of inhibition for S. aureus and E. coli were  $21 \pm 0.12$  and  $17 \pm 0.01$  mm, respectively.

# DISCUSSION

All natural silver has been recognized as a nontoxic, safe inorganic antimicrobial agent for centuries. The extracellular synthesis of silver nanoparticles had been reported numerous from bacteria, yeasts, fungi, and algae [4]. The data obtained demonstrate the extracellular synthesis of silver nanoparticles using *C. albicans*. These findings are consistent with Waghmare *et al* [3] observations that silver nanoparticles could be synthesized by *Candida* species. *Candida* is a genus of yeasts and replicate quickly, therefore are easy to manipulate genetically to make them produce more desired substances for synthesis. *C. albicans* is harmless commensals or

endosymbiosis of hosts including humans; however, when mucosal barriers are disrupted or the immune system is compromised they can invade and cause disease [14]. Moreover, Di Giacomo et al [15] used growing cells of *C. albicans* in combination with carbon nanotubes to produce stable electrically conductive bio-nanocomposite tissue materials that have been used as temperature sensing elements.

The appearance of brown color from colorless cell biomass of *C. albicans* suggested the formation of silver nanoparticles [3]. This color arises due to the surface plasmon resonance phenomenon, which monitored through the UV–visible spectrum at 420 nm, indicating formation of silver nanoparticles [3,16].

The SEM images of the particles thus obtained show different particle morphology with the size range of 20 – 80 nm. Evidence demonstrates that the morphology of nanoparticles is highly variable, and includes spherical, rod-like, decahedral, triangular, and platelet shapes. This could be due to correlation of the absorption spectrum for individual silver nanoparticles [17].

Antibacterial activity of synthesized silver nanoparticles was found to be inhibitory to both Gram negative and Gram positive bacteria. Interestingly, the synthesized silver nanoparticles have shown more antimicrobial activity against Gram positive bacteria than the Gram negative bacteria. The exact mechanism of interaction and penetration of silver nanoparticles inside the cell and their antibacterial effect remains to be identified [3,4,16,17]. Scientific evidence shows that smaller silver nanoparticles having a high definite surface area available for interaction would have more antibacterial activity than larger silver nanoparticles. Moreover, it is possible that silver nanoparticles not only interact with the surface of the membrane, but can also penetrate inside the bacteria. Furthermore, the electrostatic attraction between positively charged nanoparticles and negatively charged bacterial cell membranes could be responsible for the interactions between nanoparticles and bacterial cells [4,16,17].

The antibacterial effects of silver nanoparticles may be associated with the characteristics of certain bacterial species [4,16,17]. Our result showed the inhibition zone diameters of *S. aureus* were longer than those of *E. coli.* This implies that biosynthesized silver nanoparticles by *C. albicans* had better antibacterial activity against Gram positive bacteria than Gram negative bacteria, which could probably be

explained by their biochemical and physiological properties [18].

The hypothesized mechanisms of antibacterial activity of silver nanoparticles, mostly relate to disturbance of the integrity of the cell membrane and depletion of the levels of intracellular ATP, ROS production, and cellular uptake of Ag<sup>+</sup> ions or nanoparticles, due to membrane poration, with consequent suppression of respiratory enzymes and electron transport components and through interference with DNA functions [8,19,20]. Nada et al [5] demonstrated the antibacterial activity of silver nanoparticles synthesized from S. aureus against human pathogenic bacteria methicillin resistant S. epidermidis and Streptococcus pyogenes and against Salmonella typhi and K. pneumoniae. Similarly, El-Shanshoury et al [9] reported antibacterial activity of nanoparticles synthesized from Escherichia coli, Bacillus subtilis, and S. thermophilus against E. coli, B. subtilis, S. typhimurium, K. pneumoniae, S. aureus, Pseudomonas aeruginosa and C. albicans.

The molecular mechanisms of action of silver nanoparticles were investigated using *E. coli* as a model system by proteomic approach. Parallel proteomic investigations resulted in up-regulation in the expression of a panel of envelope and heat shock proteins, which are direct evidence of dissipation of the outer membrane of bacteria and deplete the levels of intracellular ATP [21].

#### CONCLUSION

The findings of this study demonstrate the simple, safe, cost-effective and ecofriendly preparation of silver nanoparticles using yeast *C. albicans*. The biosynthesized nanoparticles showed potential antibacterial activity against Gram-negative and Gram-positive bacteria. Thus, application of biosynthesized silver nanoparticles may lead to the development of suitable pharmaceutical and other industrial products.

# **ACKNOWLEDGEMENT**

The authors would like to thank Islamic Azad University of Yasooj for their funding of this research (no. 14-07-94-13-4).

#### REFERENCES

 Gopinath PM, Narchonai G, Dhanasekaran D, Ranjani, A, Thajuddin N. Mycosynthesis, characterization and antibacterial properties of AgNPs against multidrug

- resistant (MDR) bacterial pathogens of female infertility cases. Asian J Pharma Sci 2015; 10(2): 138–145.
- Marambio-Jones C, Hoek EMV. A review of the antibacterial effects of silver nanomaterials and potential implications for human health and the environment. J Nanopart Res 2010; 12(5): 1531–1551.
- Waghmare SR, Mulla MN, Marathe SR, Sonawane KD. Ecofriendly production of silver nanoparticles using Candida utilis and its mechanistic action against pathogenic microorganisms. 3 Biotech 2014; 5(1): 33– 38.
- Rauwel P, Küünal S, Ferdov S, Rauwel E. A review on the green synthesis of silver nanoparticles and their morphologies studied via TEM. Adv Mater Sci Eng 2015; doi.org/10.1155/2015/682749.
- Nanda A, Saravanan M. Biosynthesis of silver nanoparticles from Staphylococcus aureus and its antimicrobial activity against MRSA and MRSE. Nanomedicine 2009; 5(4): 452–456.
- Bhati-Kushwaha H, Malik CP. Biopotential of Verbesina encelioides (stem and leaf powders) in silver nanoparticle fabrication. Turkish J Biol 2013; 37(6): 645–654.
- Singh D, Rathod V, Ninganagouda S, Hiremath J, Singh AK, Mathew J. Optimization and characterization of silver nanoparticle by endophytic fungi Penicillium sp. isolated from Curcuma longa (turmeric) and application studies against MDR E. coli and S. aureus. Bioinorg Chem Appl 2014; doi.org/10.1155/2014/408021.
- 8. Rizzello L, Pompa PP. Nanosilver-based antibacterial drugs and devices: mechanisms, methodological drawbacks, and guidelines. Chem Soc Rev 2014; 43(5): 5001–1518.
- El-Shanshoury AER, ElSilk SE, Ebeid ME. Extracellular biosynthesis of silver nanoparticles using Escherichia coli ATCC 8739, Bacillus subtilis ATCC 6633, and Streptococcus thermophilus ESh1 and their antimicrobial activities. Nanotechnology 2011; doi.org/10.5402/2011/385480.
- Khodashenas B, Ghorbani HR. Evaluation of the effective factors on size and anti-bacterial properties of biosynthesized silver nanoparticles. Int J Nano Dimens 2015; 6(1): 111–127.
- Jain N, Bhargava A, Majumdar S, Tarafdar J, Panwar J. Extracellular biosynthesis and characterization of silver

- nanoparticles using Aspergillus flavus NJP08: A mechanism perspective. Nanoscale 2011; 3(2): 635–641.
- 12. Rahi DK, Parmar AS. Mycosynthesis of silver nanoparticles by an endophytic Penicillium species of Aloe vera root, evaluation of their antibacterial and antibiotic enhancing activity. Int J Nanomaterials Biostructures 2014; 4(3): 46–51.
- 13. Clinical and Laboratory Standards Institute (CLSI), "Performance standards for antimicrobial disk susceptibility tests," Approved Standard, 11th Edition, CLSI document M02-A11, 2012.
- 14. Khodavandi A, Harmal NS, Alizadeh F, Scully OJ, Sidik SHM, Othman F, Sekawi Z, Ng KP Chong PP. Comparison between allicin and fluconazole in Candida albicans biofilm inhibition and in suppression of HWP1 gene expression. Phytomedicine 2011; 19(1): 56–63.
- Di Giacomo R Maresca B, Porta A, Sabatino P, Carapella G, Neitzert H. Candida albicans/MWCNTs: A stable conductive bio-nanocomposite and its temperaturesensing properties. IEEE Transactions Nanotechnol 2013; 12(2): 111–114.
- Mie R, Samsudin MW, Din LB, Ahmad A, Ibrahim N, Adnan SNA. Synthesis of silver nanoparticles with antibacterial activity using the lichen Parmotrema praesorediosum. Int J Nanomedicine 2014; 9: 121–127.
- Mock JJ, Barbic M, Smith DR, Schultz DA, Schultz S. Shape effects in plasmon resonance of individual colloidal silver nanoparticles. J Chem Phys 2002; 116: 6755–6760.
- Umadevi M, Rani Balakrishnan T, Ramanibai R. Antimicrobial activity of silver nanoparticles prepared under an ultrasonic field. Int J Pharma Sci Nanotechnol 2011; 4(3): 1491–1496.
- Morones JR, Elechiguerra JL, Camacho A, Holt K, Kouri JB, Tapia J, Yacaman MJ. The bactericidal effect of silver nanoparticles. Nanotechnology 2005; 16(10): 2346–2353.
- 20. Li Y, Leung P, Yao L, Song QW, Newton E. Antimicrobial effect of surgical masks coated with nanoparticles. J Hosp Infect 2006; 62(1): 58–63.
- 21. Lok CN,, Ho CM, Chen R, He QY, Yu WY, Sun H, Tam PK, Chiu JF, Che CM. Proteomic analysis of the mode of antibacterial action of silver nanoparticles. J Proteome Res 2006; 5(4): 916–924.